## **PROVISIONAL ANSWER KEY**

Question 136/2024/OL

Paper Code:

Category 439/2023

Code:

Exam: Field Officer

Date of Test 26-12-2024

Department Kerala State co Op Rubber Marketing Fedn Ltd

Question1:-The sequence of reactivity for the following halogen fluorides is

 $A:-ClF_3 > BrF_5 > Clf > IF_3$ 

B:- $ClF_3 > BrF_5 < ClF > IF_3$ 

 $C:-ClF_3 < BrF_5 < ClF < IF_3$ 

 $D:-ClF_3 < BrF_5 > ClF > IF_3$ 

Correct Answer:- Option-A

Question2:-The hybridization of  $x_e$  in  $x_{eO_3}$  and geometry of  $x_{eOF_4}$  are respectively

A:-sp3d and Square Pyramidal

B:-Sp2 and Trigonal BiPyramidal

C:-Sp3d2 and Square Pyramidal

D:- $S_{p^3}$  and Square Pyramidal

Correct Answer:- Option-D

Question3:-As per molecular orbital theory, the bond order of nitrosonium ion and superoxide ion are respectively,

A:-2.50 and 1.50

B:-1.50 and 3.00

C:-3.00 and 1.50

D:-2.50 and 2.00

Correct Answer:- Option-C

Question4:-Identify the correct statements from the following:

i. Ions without noble gas configuration will have high charges at their surfaces and thus

highly polarising.

ii. The number of -OH groups present in pyrophosphoric acid and hypophosphorous acid are

different.

- iii. Tl(1) compounds are more stable than Tl(III) compounds.
- iv. The polar diagrams do not represent the total wave function  $(\boldsymbol{\psi}),$  but only the angular part

of the wave function.

A:-i, ii, iii and iv are correct

B:-ii, iii and iv are correct

C:-i, iii and iv are correct

D:-i, ii and iv are correct

Correct Answer:- Option-C

Question5:-Which of the following scientists provided experimental evidence that confirmed that electrons exhibit wave like properties as predicted by De Broglie?

A:-Germer and Davisson

B:-Albert Einstein and Heisenberg

C:-Davisson and Albert Einstein

D:-Davisson and De Broglie

Correct Answer:- Option-A

Question6:-Which one of the following statement is not correct regarding boron and its compounds?

A:-The principal oxide of boron is boric oxide

B:-Metaboric acid is H<sub>2</sub>BO<sub>2</sub>

C:-Borazine has a regular plane hexagonal structure

D:-Boron tri halides are volatile, highly reactive which show no detectable tendency to dimerise

Correct Answer:- Option-B

Question7:-Which of the following coordination compounds exhibit both ionization and linkage isomerism?

i. 
$$\left[Co(NH_3)_5NO_2\right]Cl$$

ii. 
$$\left[ Co(NH_3)_5 SCN \right] Cl$$

$$\qquad \qquad [Co(NH_3)_5Br]Cl$$

$$\text{iv.} \quad \left[ \text{Co(NH}_3)_5 H_2 O \right] C l$$

A:-All the above

B:-i and ii

C:-ii and iii

D:-iii and iv

Correct Answer:- Option-B

Question8:-The product obtained when hydrogen peroxide is added to an acidified solution of a dichromate is

$$A:=CrO(O_2)_2$$

$$B:-H_2Cr_2O_7$$

C:-CrO2

D:-CrO4

Correct Answer: - Option-A

Question9:-The sum of the first three ionization energies is maximum for which of the following lanthanides

A:- $Gd^{+3}$  and  $Yb^{+3}$ 

B:- $Eu^{+3}$  and  $Lu^{+3}$ 

C:- $Eu^{+3}$  and  $Gd^{+3}$ 

D:- $Eu^{+3}$  and  $Yb^{+3}$ 

Correct Answer:- Option-D

Question 10:- From the following complexes of Nickel, identify which are tetrahedral?

A:- $[Ni(CO)_4]$  and  $K_2[Ni(CN)_4]$ 

 $B:-[Ni(Br)(Pph_3)_3]$  and  $K_2[Ni(CN)_4]$ 

 $C:-K_2[Ni(Cl)_4]$  and  $K_2[Ni(CN)_4]$ 

 $D:-[Ni(Br)(Pph_3)_3]$  and  $[Ni(CO)_4]$ 

Correct Answer: - Option-D

Question11:-Reduction of potassium permanganate with aqueous Na2SO3 produces

A:-Trioxomanganate (VI)

B:-Managanous oxide (IV)

C:-Tetraoxomanganate (V)

D:-Dioxomanganate (V)

Correct Answer:- Option-C

Question12:-Biodegradable non persistent Pesticide which is considered as a neurotoxin that causes immediate paralysis to insects

A:-Mirex

B:-Endrin

C:-Heptachlor

D:-Pyrethrin

Correct Answer:- Option-D

Question13:-The Resin Identification Codes or the "Chasing arrows" triangle with recycling codes 3 & 4 are for the following

A:-Poly Vinyl Chloride and Low Density Poly Ethylene

B:-Poly Propylene and Poly Vinyl Chloride

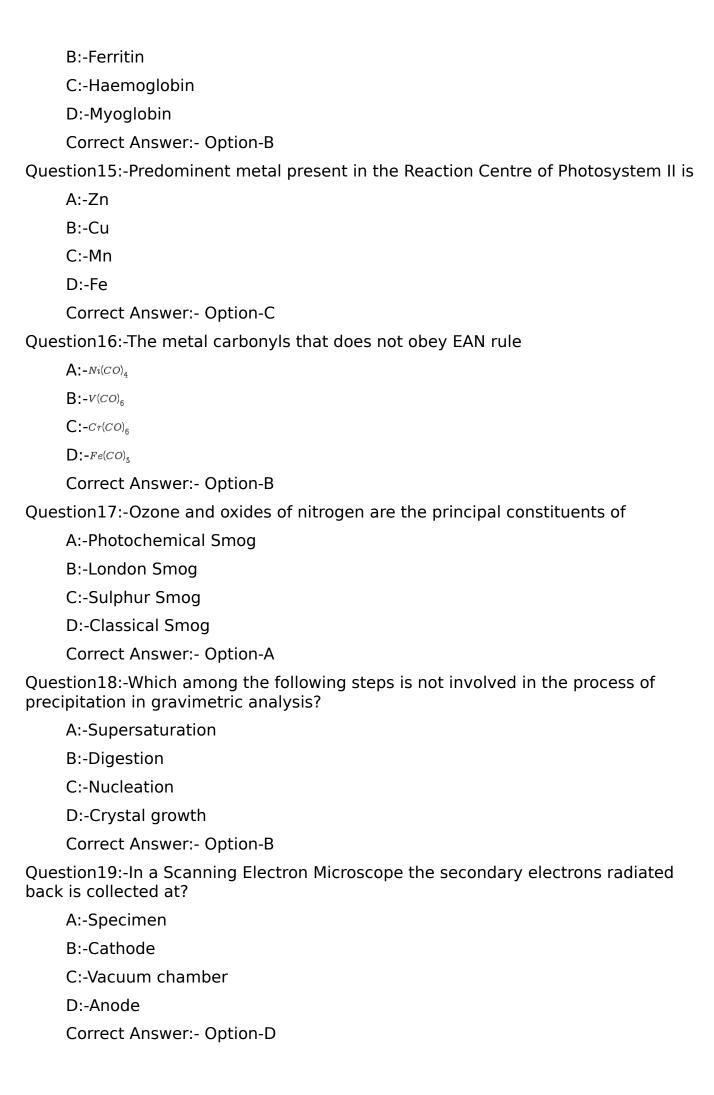
C:-High Density Poly Ethylene and Low Density Poly Ethylene

D:-Low Density Poly Ethylene and High Density Poly Ethylene

Correct Answer:- Option-A

Question14:-Storage of iron in the body is carried out by

A:-Cytochrome P-450



Question20:-Which among the following stationary phases act as a strong acidic cation exchange resin

A:-Quarternary Ammonium Polystyrene

B:-Sulphonated Polystyrene

C:-Polyamine Polystyrene

D:-Carboxylic Polymethyl Methacrylate

Correct Answer:- Option-B

Question21:-Which of the following compounds show Norrish Type II reaction?

A:-Propanone

B:-3-Pentanone

C:-4,4 Dimethyl 2-Pentanone

D:-2, 4 Dimethyl 3-Pentanone

Correct Answer:- Option-C

Question22:-Arrange the following compounds in the order of decreasing reactivity towards electrophilic substitution reaction.

i. Phenol

ii. Toluene

iii. Benzaldehyde

iv. Aniline

A:-iv>i>ii>iii

B:-i>iii>iv>ii

C:-iii>iv>ii>i

D:-i>iv>iii>ii

Correct Answer:- Option-A

Question23:-Select the incorrect statement among the following for  $S_{N^2}$  reaction

A:-Change of nucleophile alter the rate of the reaction

B:-Stronger the nucleophile faster will be the rate

C:-The rate of the reaction will be faster if the leaving group is less basic

D:-The rate of reaction is independent of the concentration of nucleophile

Correct Answer:- Option-D

Question24:-How many of the following species are aromatic in nature?



A:-2

B:-3

C:-4

D:-5

Correct Answer:- Option-B

Question25:-Among the following Z-Isomer is:

H<sub>3</sub>C \_ \_ |

B:- CI

C:- Br

H<sub>3</sub>C

Correct Answer:- Option-C

Question26:-Which among the following compound is formed when Acetone reacts with Methyl Magnesium bromide followed by hydrolysis?

A:-tert-Butyl alcohol

B:-n-butyl alcohol

C:-Iso propyl alcohol

D:-n-propyl alcohol

Correct Answer:- Option-A

Question27:-The following reaction is an example of :

C<sub>6</sub>H<sub>5</sub>CHO+CH<sub>3</sub>CHO Dil NaOH → C<sub>6</sub>H<sub>5</sub>CH=CHCHO +H<sub>2</sub>O

A:-Benzoin condensation

**B:-Claisen Schmidt Reaction** 

C:-Reformatsky Reaction

D:-Perkin Condensation

Correct Answer:- Option-B

Question28:-Oxidation of glucose with nitric acid in the presence of  $FeSO_4$  and  $V_2O_5$  as catalyst gives

A:-Oxalic acid

B:-Adipic acid

C:-Citric acid

D:-Anthranilic acid

Correct Answer:- Option-A

Question29:-Reaction of Sodium phenoxide with Cl4 in alkaline medium gives

A:-Salicylaldehyde

**B:-Salicylic Acid** 

C:-Aspirin

D:-Salol

Correct Answer:- Option-B

Question 30:-Which of the following compounds is most reactive with anhydrous  $z_{nCl_2}$  and Con. HCl?

A:-2-Methyl propanol

B:-1-Propanol

C:-2-Propanol

D:-2-Methyl-2-propanol

Correct Answer:- Option-D

Question31:-Which among the following compounds undergo Cannizzaro reaction?

A:-Acetone

B:-Acetaldehyde

C:-Butanone

D:-Trimethyl Acetaldehyde

Correct Answer:- Option-D

Question32:-In which of the following terpenes does the isoprene rule not hold?

A:-Limonene

B:-Menthol

C:-Squalene

D:-Myrcene

Correct Answer:- Option-C

Question33:-Which of the following types of plastic undergoes irreversible curing upon heating and cannot be remoulded?

A:-Polvethylene

B:-Polystyrene

C:-Polyvinyl chloride

D:-Polyurethane

Correct Answer:- Option-D

Question34:-In which of the following reactions does diazonium chloride play a crucial role in organic synthesis?

A:-Buchwald-Hartwig amination

**B:-Sandmeyer reaction** 

C:-Wittig reaction

D:-Mannich reaction

Correct Answer:- Option-B

Question35:-Which of the following compounds will give a positive carbylamine test?

A:-N,N- Diethyl aniline

B:-P- Ethyl Benzylamine

C:-Triethylamine

D:-Diphenylamine

Correct Answer:- Option-B

Question36:-Identify the following plane



A:-(011) plane in BCC unit cell

B:-(100) plane in FCC unit cell

C:-(100) plane in BCC unit cell

D:-(110) plane in FCC unit cell

Correct Answer:- Option-C

Question 37:-Calculate the compressibility factor for  $N_2$ , if two mole of it occupies 0.2 L at 100 K and 82.1 atm.

A:-10

B:-1

C:-50

D:-5

Correct Answer:- Option-B

Question38:-Which of the following describes a Schottky defect in a crystal lattice?

A:-A vacancy created by the absence of a cation and an anion in the lattice structure.

- B:-A defect formed due to the presence of an extra ion in the crystal lattice.
- C:-An imperfection caused by the substitution of one type of ion with another type of ion in the lattice.
  - D:-A defect resulting from the displacement of atoms due to external pressure Correct Answer:- Option-A

Question39:-Which of the following correctly describes the types of statistics used to describe different systems of particles?

- A:-Maxwell-Boltzmann statistics describe indistinguishable particles with no restrictions on the number of particles per state.
- B:-Bose-Einstein statistics describe indistinguishable particles that obey the Pauli exclusion principle.
- C:-Fermi-Dirac statistics describe indistinguishable particles with no restrictions on the number of particles per state.
- D:-Fermi-Dirac statistics describe indistinguishable particles that obey the Pauli exclusion principle.

Correct Answer:- Option-D

Question 40:-Which of the following statements correctly describes the Joule-Thomson effect?

- A:-The Joule-Thomson effect describes the temperature change of a real gas when it is expanded or compressed adiabatically at constant internal energy.
- B:-The Joule-Thomson effect describes the temperature change of an ideal gas when it is expanded or compressed adiabatically at constant pressure.
- C:-The Joule-Thomson effect describes the temperature change of a real gas when it is expanded or compressed adiabatically at constant enthalpy.
- D:-The Joule-Thomson effect describes the pressure change of a real gas when it is expanded or compressed isothermally at constant enthalpy.

Correct Answer:- Option-C

Question41:-Molar conductivity at infinite dilution of weak electrolytes can be determined by

A:-Plotting concentration vs conductance graph

B:-Kohlrausch's law

C:-Both (1) and (2)

D:-Potentiometric titrations

Correct Answer:- Option-B

Question 42:- The amount of Al deposited on passing 96500 Coulombs of charge through a solution of  $AI(NO_3)_3$  is

A:-9

B:-27

C:-18

D:-10

Correct Answer: - Option-A

Question43:-In which of the following reactions, the value of Kp will be equal to Kc?

A:-
$$N_{2(g)}$$
 +  $3H_{2(g)} \rightleftharpoons 2NH_{3(g)}$ 

B:-

$$N_{2(g)}$$
 +  $O_{2(g)} \rightleftharpoons 2NO_g$ 

$$C:-2SO_{2(g)}$$
 +  $O_{2(g)} \implies 2SO_{3(g)}$ 

$$D:$$
  $-CaCO_{3(s)} \implies CaO_{(s)} + CO_{2(g)}$ 

Correct Answer:- Option-B

Question44:-Calculate the overall order of the bimolecular reaction based on the following data.

 $A + B \rightarrow Product$ 

Con. A	Conc. B	Rate of the reaction
0.01M	0.01M	$1 \times 10^{3}$
0.02M	0.01M	$4 \times 10^{3}$
0.01M	0.02M	$2 \times 10^{3}$

A:-0

B:-1

C:-2

D:-3

Correct Answer:- Option-D

Question45:-The pH of a solution is equal to its pKa value, when

A:-Acid is consumed completely

B:-Base is consumed completely

C:-The concentration of conjugate base and conjugate acid are equal

D:-All of the above

Correct Answer:- Option-C

Question46:-Which of the following molecules will have a center of symmetry?

A:-Water

B:-Ammonia

C:-Hydrogen

D:-Boron trifluoride

Correct Answer:- Option-C

Question47:-Which of the following factors will affect the Chemical shift values in NMR spectroscopy?

A:-Size of the atom

B:-Electron density around the central atom

C:-Applied Magnetic field

D:-Solvent used while recording spectrum

Correct Answer:- Option-B

Question 48:-What is the result of applying operator  $A = \frac{d^2}{dx^2}$  to the function  $f(x) = 10x^3$ ?

A:-30x<sup>2</sup>

B:-30x3

C:-60x2

D:-60x

Correct Answer:- Option-D

Question49:-What is the energy obtained for a particle in a 3D box?

$$A:-E = \frac{h^2\pi^2n^2}{8mn}$$

$$B: -E = \frac{h^2}{2\pi m} \left( \frac{n_x^2}{a} \right)$$

$$C:=E=\frac{h^2\pi^2}{8mabc}$$

$$D: -E = \frac{h^2}{8m} \left( \frac{n_x^2}{a} + \frac{n_y^2}{b} + \frac{n_z^2}{c} \right)$$

Correct Answer:- Option-D

Question 50:- Why are stokes lines more intense than anti-stokes lines in Raman effect?

A:-Ground and excited states are equally populated

B:-Excited state is more populated than ground state

C:-Ground state is more populated than excited state

D:-None of the above

Correct Answer:- Option-C

Question51:-Choose the CORRECT answer.

- i. The cambium ring in roots is partly primary and partly secondary in origin.
- ii. Medullary rays are less distinct in sapwood.
- iii. Tracheid parenchyma contribute 90-95% in hardwood.
- iv. Vascular cambium and cork cambium are lateral meristems.

A:-True, True, False, False

B:-True, False, True, False

C:-False, False, True, True

D:-False, True, False, True

Correct Answer:- Option-D

Question52:-The position of phloem in the vertical section of a typical dicot leaf is

A:-Facing the upper epidermis.

B:-Facing the lower epidermis.

C:-Facing lateral sides.

D:-All around the xylem.

Correct Answer:- Option-A

Question53:-Choose the CORRECT answer.  Nucleated cells of phloem:; Living cells of xylem:	
A:-Sieve tubes; Tracheids	
B:-Companion cells; Ray & Parenchyma	
C:-Phloem Fibres; Xylem fibres	
D:-Phloem parenchyma; Vessels	
Correct Answer:- Option-B	
Question54: cannot bring genetically same type of pollens on stigma.	
A:-Autogamy	
B:-Cleistogamy	
C:-Geitonogamy	
D:-Xenogamy	
Correct Answer:- Option-D	
Question55:-Some statements regarding the types of embryo sacs in angiosper are given below. Choose the INCORRECT one. i. In <i>polygonum</i> type, the nucleus divides thrice to form the eight nucleated embryo sac.	
<ul><li>ii. Allium type develops from the micropylar dyad.</li><li>iii. Endymion type develops from the chalazal dyad.</li><li>iv. Nuclear fusion occurs in Peperomia type.</li></ul>	
A:-i & ii	
B:-ii & iv	
C:-i, iii, iv	
D:-ii, iii, iv	
Correct Answer:- Option-D	
Question56:-The most suitable preparation to study the meiotic behavior of chromosomes is	
A:-Maceration	
B:-Smear	
C:-Squash	
D:-Whole mount	
Correct Answer:- Option-B	
Question57:-Determine the mean of the median and mode of the data given below. 5, 10, 3, 6, 4, 8, 9, 3, 15, 2, 9, 4, 19, 11, 4	
A:-5	
B:-9	
C:-10	
D:-12	
Correct Answer:- Option-A	

Question58:-Choose the Correct statement from the following:

- Thin layer chromatography is a type of partition chromatography, while paper chromatography is a type of adsorption chromatography.
- Ion exchange chromatography is affinity chromatography. ii.
- In SDS-PAGE, the proteins are separated according to their electrophoretic mobility.
- iv. Electrophoresis can be used for the separation of lipids.

A:-i and ii

B:-i and iv

C:-ii and iii

D:-ii and iv

Correct Answer:- Option-C

Question59:-Match the following and choose the correct answer:

- Aecidiospores a. *Agaricus*Ascospores b. *Peziza* i.
- ii. Ascospores
- iii. Basidiospores
- b. *Peziza* c. *Penicillium*

iv. Conidia

d. Puccinia

A:-i-a, ii-c, iii-d, iv-b

B:-i-b. ii-d. iii-c. iv-a

C:-i-c, ii-a, iii-b, iv-d

D:-i-d, ii-b, iii-a, iv-c

Correct Answer:- Option-D

Question60:-Choose the Correct answer:

\_\_\_\_\_: Early blight of Potato and Tomato;

: Ergot of rye and grasses.

: Late blight of Potato and Tomato;

: Tikka disease of groundnut.

A:-Rhizopus stolonifer, Puccinia graminis, Fusarium oxysporum, Phytophthora infestans.

B:-Alternaria solani, Claviceps purpurea, Phytophthora infestans, Cercospora personata.

C:-Phytophthora infestans, Puccinia graminis, Alternaria solani, Fusarium oxysporum.

D:-Phytophthora infestans, Claviceps purpurea, Alternaria solani, Rhizopus stolonifer.

Correct Answer:- Option-B

Question61:-Lycopodium clavatum: "Club moss"; \_\_\_\_\_: "Irish moss".

A:-Chondrus

B:-Funaria

C:-Gracilaria

D:-Sphagnum Correct Answer:- Option-A Question62:-What is Protonema? A:-Young gametophyte of Funaria B:-Gametophyte of Pteris C:-Cup shaped vegetative structure of Funaria D:-Sterile appendage in anthridial branch of Funaria Correct Answer:- Option-A Question63:-Stomium and Annulus are parts of A:-Sporogonium of Riccia B:-Capsule of Funaria C:-Sporangium of Pteris D:-Antheridium of Pteris Correct Answer:- Option-C Question64:-Function of transfusion tissue in Cycas A:-Storage **B:-Conduction** C:-Resin secretion D:-Hosting algal cells Correct Answer:- Option-B Question65:-Veterinary drug which caused massive death of vultures in South Asia A:-Ampicillin B:-Paracetamol C:-Aspirin D:-Diclofenac Correct Answer:- Option-D Question66:-Chemical which causes Blue Baby Syndrome A:-Mercury B:-Cadmium C:-Carbonate

D:-Nitrate

Question67:-Fly ash is

Correct Answer:- Option-D

A:-Waste from incinerators

C:-Ash from industrial furnace

B:-Waste from thermal power plant

D:-Ash from Petroleum

Correct Answer:- Option-B

Question68:-Black Dammer is obtained from

A:-Vateria indica

B:-Strychnos nux vomica

C:-Canarium strictum

D:-Pinus roxburghii

Correct Answer:- Option-C

Question69:-Cyathium inflorescence is a character of the family

A:-Apiaceae

B:-Asteraceae

C:-Podostemaceae

D:-Euphorbiaceae

Correct Answer:- Option-D

Question70:-Lemon is an example for

A:-Drupe

B:-Hesperidium

C:-Berry

D:-Pome

Correct Answer:- Option-B

Question71:-A fibre yielding plant

A:-Rubus fruticosus

B:-Melothria pendula

C:-Cassia angustifolia

D:-Linum usitatissimum

Correct Answer:- Option-D

Question72:-Gymnosperms dont have

A:-Tracheids

B:-Sclereids

C:-Vessels

D:-Companion cells

Correct Answer:- Option-C

Question73:-The specific functions of the plasma membrane, including selective transport of molecules and cell-cell recognition, are carried out by

A:-Phosphatidylcholine

**B:-Membrane Proteins** 

- C:-Phosphatidylserine
- D:-Sphingomyelin

Correct Answer:- Option-B

Question74:-Which of the following crosses can be used to identify whether an organism exhibiting a dominant trait is homozygous or heterozygous for a specific allele?

- A:-Testcross
- **B:-Outcross**
- C:-Both (1) and (2)
- D:-None of the above

Correct Answer:- Option-A

Question75:-Name the antigens present in O group blood

- A:-Antigen A
- B:-Antigen B
- C:-Both antigens A and B
- D:-No antigens

Correct Answer: - Option-D

Question 76:- As per Chargaff's rule, which among the following statements are correct:

- i. Adenine can pair with Guanine
- ii. Cytosine can pair with Guanine
- iii. Adenine can pair with Thymine
- iv. Cytosine can pair with Thymine
  - A:-Statements i and ii
  - B:-Statements ii and iii
  - C:-Statements i and iv
  - D:-Statement i

Correct Answer:- Option-B

Question77:-The concept of 'Natural selection and survival of the fittest' is the major thrust area in

- A:-Lamarckism
- B:-Neo-Darwinism
- C:-Darwinism
- D:-Mutation Theory

Correct Answer:- Option-C

Question 78:-Which of the following statements are true with regard to the phenomenon of guttation through hydathodes?

- i. Guttation is mostly driven by root pressure.
- ii. Guttation can happen when the stomata are closed.
- iii. Guttation can happen when the humidity is high.

- iv. Guttation is used to detoxify plant tissues by exporting excessive salts or molecules.
  - A:-Statements i and ii only
  - B:-Statements ii and iii only
  - C:-Statements iii and iv only
  - D:-All the four statements are correct

Correct Answer:- Option-D

Question79:-Light reaction of photosynthesis happens in

- A:-Grana of chloroplast
- B:-Stroma of chloroplast
- C:-Chloroplast membrane
- D:-None of the above

Correct Answer:- Option-A

Question80:-Which among the following will be the respiratory substrate if the Respiratory Quotient is 1?

- A:-Protein
- B:-Carbohydrate
- C:-Lipids
- D:-Organic acids

Correct Answer:- Option-B

Question81:-Which one of the following cannot promote nitrogen fixation in its free-living state?

- A:-Anabaena
- B:-Azotobacter
- C:-Kelbsiella
- D:-Clostridium

Correct Answer:- Option-A

Question82:-Which among the following is not an aliphatic amino acid?

- A:-Leucine
- B:-Alanine
- C:-Glycine
- D:-Phenylalanine

Correct Answer:- Option-D

Question83:-If 'C' is a cybrid of 'A' and 'B', then

A:-The nucleus of 'C' will be that of earlier 'A' or 'B' and the cytoplasm will be a hybrid of 'A' and 'B'

B:-The nucleus of 'C' will be a hybrid of 'A' and 'B' and the cytoplasm will be that of earlier 'A' or 'B'

C:-The nucleus and cytoplasm of 'C' will be hybrid of 'A' and 'B'

D:-None of the above

Correct Answer:- Option-A

Question84:-Type II Restriction endonuclease enzymes will

A:-Cleave at sites away from the recognition site

B:-Cleave within or at short specific distances from the recognition site

C:-Cleave at sites 25-27 bp from the recognition site

D:-Cleave close to or within the recognition sequence

Correct Answer:- Option-B

Question85:-Which among the following is not an advantage in RAPD technique?

A:-Does not require any specific knowledge of the DNA sequence of the target organism

B:-All RAPD markers are dominant

C:-Both (1) and (2)

D:-None of the above

Correct Answer:- Option-B

Question86:-Which among the following is not involved in the production of Single Cell Proteins?

A:-Oedogonium abbreviatum

B:-Spirulina maxima

C:-Chlorella pyrenoidosa

D:-Scenedesmus acutus

Correct Answer: - Option-A

Question87:-Which is the estimated size of human genome in base pairs?

A:-12 million

B:-97 million

C:-3 billion

D:-1 billion

Correct Answer:- Option-C

Question88:-Identify the method which is not a vegetative propagation method.

A:-Budding

B:-Hybridization

C:-Layering

D:-Grafting

Correct Answer:- Option-B

Question89:-Sharbati Sonora is a mutated Mexican dwarf variety of the crop

A:-Rice

B:-Sugarcane

C:-Wheat

D:-Green gram

Correct Answer: - Option-C

Question90:-Raphanobrassica is a hybrid between

A:-Radish and Cabbage

B:-Radish and Carrot

C:-Radish and Tomato

D:-Radish and Potato

Correct Answer:- Option-A

Question91:-Which of the following best defines polyclonal seed germination in rubber cultivation?

A:-Germination of seeds produced through self-pollination between genetically identical rubber trees.

B:-Propagation of rubber trees from seeds resulting from controlled cross-pollination between genetically diverse parent trees.

C:-Cultivation of rubber trees from seeds generated through gamma radiationinduced mutations.

D:-Selection of seeds based on specific genetic traits to enhance latex yield in rubber plants.

Correct Answer:- Option-B

Question92:-Which of the following is considered the best vegetative propagation method in rubber cultivation?

A:-Cutting

B:-Layering

C:-Bud grafting

D:-Approach grafting

Correct Answer:- Option-C

Question 93:-Which of the following best describes the function of the latex vessel ring in rubber trees?

A:-Facilitating water transport from roots to leaves.

B:-Serving as a storage site for latex production.

C:-Regulating the flow of latex during tapping.

D:-Protecting the inner bark from damage.

Correct Answer:- Option-C

Question94:-What does IRRDB stand for in the context of rubber cultivation?

A:-International Rubber Research and Development Board

B:-Institute of Rubber Research and Development Bureau

C:-International Rubber Research and Development Bureau

D:-Institute of Rubber Research and Development Board

Correct Answer:- Option-A

Question 95:-What is the common intermediate compound produced during the synthesis of Hevea rubber?

A:-Isopentenyl diphosphate (IDP)

B:-Squalene

C:-Mevalonic acid

D:-Farnesyl pyrophosphate (FPP)

Correct Answer:- Option-A

Question 96:-Which of the following types of tapping knives is most commonly used for rubber tapping?

A:-Budding knife

B:-Machete

C:-Jebong knife

D:-Bush knife

Correct Answer:- Option-C

Question 97:-What does TPD stand for in the context of rubber tapping?

A:-Tapping Procedure Development

**B:-Total Production Data** 

C:-Tapping Panel Dryness

D:-Tree Productivity Duration

Correct Answer:- Option-C

Question 98:-Which of the following fungi is primarily responsible for causing powdery mildew in rubber trees?

A:-Colletotrichum gloeosporioides

B:-Oidium heveae

C:-Phytophthora palmivora

D:-Fusarium oxysporum

Correct Answer:- Option-B

Question99:-Besides latex and timber, what other product can be harvested from rubber trees to generate additional income?

A:-Rubber tree resin

B:-Rubber tree bark for medicinal use

C:-Honey from rubber tree flowers

D:-Rubber leaves for composting

Correct Answer:- Option-C

Question100:-What are root trainers in the context of rubber biology?

A:-Devices used for tapping latex from rubber trees

B:-Structures designed to support the growth of rubber tree roots

C:-Tools for applying fertilizers to rubber trees

D:-Containers used for propagating rubber tree seedlings

Correct Answer:- Option-D